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A New Synthetic Route to Macrocycles : Synthesis of Large Ring Enaminolactones

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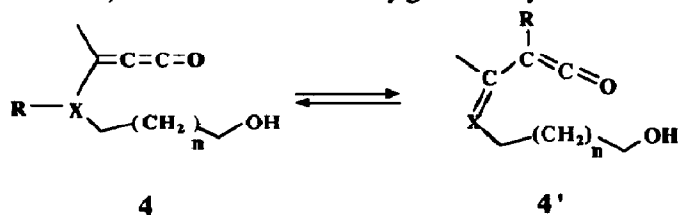
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Abstract : *Macrocyclic enaminolactones 5 are prepared in three steps from commercially available chloroalcohols 1. The cyclization step is performed by intramolecular nucleophilic addition of the hydroxy group to an aminomethyleneketene generated by thermolysis.*

Macrocycles and particularly macrolides which sometimes contain an α,β -unsaturated lactone moiety¹ have attracted interest over a number of years and synthetic methods towards these large ring cycles have been described.²⁻⁴ Macrolides generally occur in nature and many of them have biological activity such as antibiotic,⁵ antiviral⁶⁻⁸ or cytotoxic properties.⁹

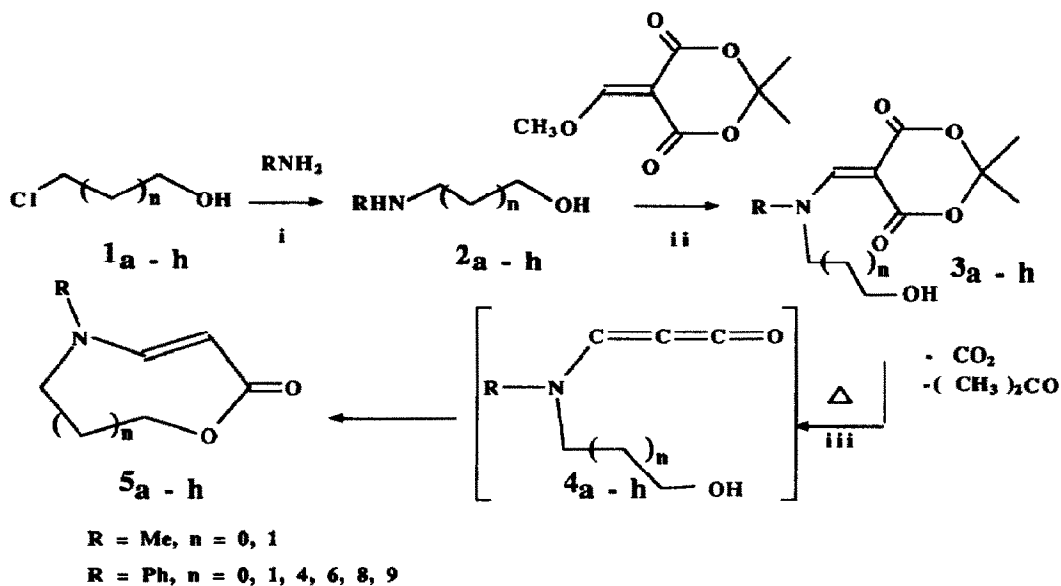
The synthesis of macrocycles can be carried out from either linear or cyclic precursors : in the past few years this has been done by increasing the size of smaller cycles using silylated compounds,¹⁰ and by Wittig¹¹ or Ullmann reactions.¹² Enzymatic Baeyer-Villiger reaction have also been applied to the formation of seven membered lactones.¹³

More recently, Quinkert and coworkers have obtained optically active macrocyclic lactones or lactams by photo-lactonization or lactamization of ortho-quinol acetates :^{14,15} the cyclization was based on an intramolecular nucleophilic attack of hydroxy or amino group respectively on the photochemically generated dieneketene intermediate which is a vinylogue of 4' (X = CH). Owing to the present interest in macrolides, we decided to investigate such a strategy and we present here our results concerning a general synthesis of a new class of macrocycles (β -enaminolactones) based on the reactivity of methyleneketenes.¹⁶ The key step involves the cyclization of ω -hydroxyalkylamino methyleneketenes 4 (X = N) which are formally the tautomeric forms of iminoketenes 4' (X = N when R = H). Intermediates 4 are easily generated by thermal decomposition¹⁷ of the



corresponding Meldrum acid derivatives (5-(ω -hydroxyalkylaminomethylene)-2,2-dimethyl-1,3-dioxane-4,6-dione 3). This new methodology supposes the preliminary synthesis of secondary alkylamines 2 bearing a hydroxygroup at the ω - position.

Aminoalcohols **2** were prepared from commercially available chloroalcohols **1** : a large excess of amine (10 or 15 equivalents) was slowly added to a solution of methylamine in ethanol (**2 a, b**, R = CH₃) or aniline in toluene (**2 c-h**, R = Ph), and the mixture was stirred at 25 ° or 110 ° C respectively for 15 or 24 hours. After

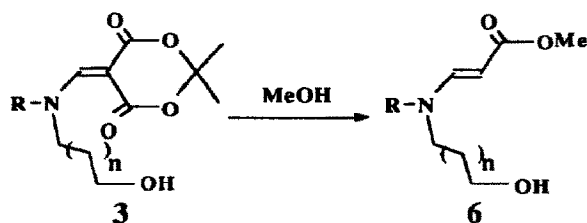


i = 10 eq of aniline or methylamine per mole of **1** in toluene ;

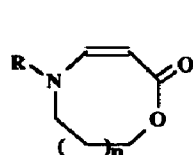
ii = 1 eq of methoxymethylene Meldrum acid derivative per mole of **2**, acetonitrile as solvent ;

iii = thermolysis : temperature range, 400 ° - 500° C ; pressure : 16 torr.

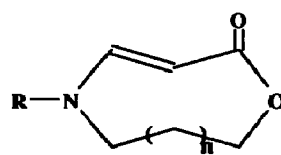
extraction, aminoalcohols **2** were isolated by distillation or recrystallization, but could also be immediately transformed into Meldrum acid derivatives **3 a-h** in good yields without previous purification . In the second step, substitution of the alkoxy group of 5-methoxymethylene-2,2-dimethyl-1,3-dioxane-4, 6-dione¹⁸ by the amino group of **2** was performed under reflux in acetonitrile for 8 hours. It should be mentioned that when the mixture was maintained much longer at 80 ° C , the Meldrum's acid ring could be cleaved by methanol formed during the reaction and thus the corresponding enamoesters **6** were exclusively obtained. Similar monodecarboxylating transesterifications leading to β -enamoesters have already been reported in basic or acid conditions ;^{19, 20} these have also been reported in a neutral medium.²¹



The cyclization was achieved by thermolysis of Meldrum acid derivatives **3 a-h** between 400 and 500 °C under flow conditions¹⁶: in all cases, enaminolactones **5 a-h** were obtained. These results can be rationalized by assuming the formation of the methyleneketene intermediates **4 a-h** generated from **3 a-h** by thermal extrusion of carbon dioxide and acetone.¹⁷ Intramolecular nucleophilic addition of the terminal hydroxy group to the ketene moiety of the heterocumulene affords medium-ring lactones **5 a-d** with only the Z double bond configuration ($n = 0, 1$; R = CH₃ or Ph, $J_{2,3} = 9,5-10$ Hz) whereas for the larger ring derivatives **5 e-h** ($n = 4, 6, 8, 9$, R = Ph) the ethylenic bond is exclusively E ($J_{2,3} = 13-13,5$ Hz).



5 a,b R: Me n: 0,1
5 c,d R: Ph n: 0,1



5 e-h R: Ph n: 4,6,8,9

Isolated yields of 2, 3 and 5

R	Me	Me	Ph	Ph	Ph	Ph	Ph	Ph
n	0	1	0	1	4	6	8	9
(%) 2	(1)	40	(1)	64	89	98	92	94
(%) 3	95	90	92	57	81	99	97	97
(%) 5	60	35	57	56	22	36	75	54

(1): commercially available

In summary, this three step sequence provides a stereospecific synthetic access to large sized enaminolactones, a novel class of macrocycles, with relatively good yields.

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References and Notes :

1. Omura, S., *Macrolide antibiotics Chemistry, Biology and Practice*; Academic Press, New- York, 1984.
2. Nicolaou, K. C., *Tetrahedron*, 1977, 33, 683 - 710.
3. Masamune, S.; Bates G. S.; Corcoran, J. W., *Angew. Chem. Int. Ed. Engl.*, 1977, 16, 585 - 607.
4. Back, T. G., *Tetrahedron*, 1977, 33, 3041 - 3059.
5. a. Brockmann, H; Henkel, W., *Chem. Ber.* 1951, 84, 284 - 288.

- b. Boschelli, D. ; Takemasa, T. ; Nishitani, Y. ; Masamune S., *Tetrahedron Lett.*, **1985**, *26*, 5239 - 5242.
- c. Kennedy, R. M. ; Abiko, A. ; Masamune, S., *ibid*; **1988**, *29*, 447 - 450.
- d. Nicolaou, K. C. ; Chakraborty, T. K. ; Ogawa, Y. ; Daines, R. A. ; Simpkins, N. S.; Furst, G. T., *J. Am. Chem. Soc.*, **1988**, *110*, 4660 - 4672.
- e. Evans, D. A. ; Calter, M. A., *Tetrahedron Lett.* , **1993**, *34*, 6871 - 6874.
- f. Rouch, W. R. ; Bannister, T. D., *Tetrahedron Lett.* , **1992**, *33*, 3587 - 3590.
- g. Sekiguchi, J. ; Kuroda, H. ; Yamada, Y. ; Okada, H., *Tetrahedron Lett.*, **1985**, *26*, 2341 - 2342.
6. Higa, T. ; Tanaka, J. ; Komesu, M. ; Garcia - Gravalos, D. ; Puentes, J. L. F. ; Bernardinelli, G. ; Jefford, C. W., *J. Am. Chem. Soc.*, **1992**, *114*, 7587 -7588. .
7. Kobayashi, J. ; Ishibashi, M. ; Murayama, T. ; Takamatsu, M. ; Iwamura, M. ; Ohizumi, Y. ; Sasaki, T., *J. Org. Chem.*, **1990**, *55* , 3421 - 3423.
8. Pettit, G.R. ; Singh, S. B. ; Niven, M. L., *J. Am. Chem. Soc.*, **1988**, *110*, 8539 - 8540.
9. Kobayashi, J. ; Shigemori, H. ; Ishibashi, M. ; Yamasu, T. ; Hirota H. ; Sasaki, T., *J. Org. Chem.*, **1991**, *56*, 5221 - 5224 and references cited therein.
10. Fouque, E. ; Rousseau, G. ; Seyden-Penne, J., *J. Org. Chem*, **1990**, *55*, 4807 - 4817.
11. Le Floch, Y. ; Yvergnaux, F. ; Toupet, L.; Gree, R. , *Bull. Soc. Chim. Fr.*, **1991**, *128*, 742 - 759.
12. Boger, D. L. ; Sakya, S. M. ; Johannes, D. , *J. Org. Chem.*, **1991**, *56*, 4204 - 4207.
13. Walsh, C. T. ; Chen, Y. C. J., *Angew. Chem. Int. Ed. Engl.*, **1988**, *27*, 333 - 343. .
14. Quinkert, G. ; Billhardt, U. M. ; Jakob, H. ; Fischer, G. ; Glenneberg, J. ; Nagler, P. ; Autze, V. ; Heim, N. ; Wacker, M. ; Schwalbe, Th. ; Kurth, Y.; Bats, J. W. ; Dürner, G., Zimmermann, G. ; Kessler, H. , *Helv. Chim. Acta* , **1987**, *70*, 771 - 861. .
15. Quinkert, G. ; Nestler, H. P. ; Schumacher, B. ; del Grosso, M. ; Dürner, G. ; Bats, J. W., *Tetrahedron Lett.*, **1992**, *33*, 1977 - 1980.
16. Grandjean, D. ; Dhimane, H. ; Pommelet, J. C. ; Chucho, J. , *Bull. Soc. Chim. Fr.*, **1989**, 657 - 660.
17. a. Lorencak, P. ; Pommelet, J. C. ; Chucho, J. ; Wentrup, C., *J. Chem. Soc., Chem. Comm.* , **1986**, 369 - 370. .
- b. Ben Cheikh, A. ; Dhimane, H. ; Pommelet, J. C. ; Chucho, J., *Tetrahedron Lett.*, **1988**, *29*, 5919 - 5922. .
18. The 5-methoxymethylene-2,2-dimethyl-1,3-dioxane-4,6-dione was easily prepared by a standard method: Bihlmayer, G. A. ; Derflinger, G. ; Derkosch, J. ; Polansky, O.E., *Monatsh. Chem.* , **1967**, *98*, 564. - 578.
19. a. Célérier, J. P. ; Deloisy, E. ; Kapron, P ; Lhommet, G. ; Maitte, P., *Synthesis*, **1981**, 130 - 133. .
- b. Célérier, J.P. ; Lhommet, G. ; Maitte, P., *Tetrahedron Lett.*, **1981**, *22*, 963 - 964.
- c. Célérier, J. P. ; Richaud, M. G. ; Lhommet, G., *Synthesis*, **1983**, 195. - 197.
20. Corey, E. J., *J. Am. Chem. Soc.* , **1952**, *74*, 5897 - 5905. .
21. Oikawa, Y. ; Sugano, K. ; Yonemitsu, O. , *J. Org. Chem.*, **1978**, *43*, 2087 - 2088. .

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